Covalent Bonding Chart 1 AP Chemistry

| Molecule and formula | Dot structure, resonance structures if applicable, formal charges | Shape name, 3D drawing, angles | Polar bonds? | Polar molecule? | IMF |
|---|---|--------------------------------|-----------------|--------------------|-----|
| Boron tribromide | | | | | |
| Carbon dioxide | | | | | |
| Phosphorus trichloride | | | | | |
| Dichloromethane (CH ₂ Cl ₂) | | | | | |

| Water | | | |
|---------------------------|--|--|--|
| Phosphorus pentaiodide | | | |
| Dihydrogen monosulfide | | | |
| Hydrochloric acid | | | |
| Carbon tetrabromide | | | |

| Thiocyanate ion | | | |
|-------------------------------------|--|--|--|
| Sulfate ion | | | |
| OPBr ₃ (P is central) | | | |
| ocs | | | |
| Xenon tetrachloride | | | |

| Triazine (C ₃ H ₃ N ₃) Ring with alternating C and N | | | |
|---|--|--|--|
| Chlorine monoxide | | | |
| Propene (C ₃ H ₆) | | | |
| Dinitrogen monoxide (central atom is N) | | | |
| Iodine trichloride | | | |