

# NAMING COMPOUNDS (MIXED)

Name \_\_\_\_\_

Name the following compounds.

1. NaCl \_\_\_\_\_
2. MnS \_\_\_\_\_
3.  $K_2O$  \_\_\_\_\_
4.  $CuBr_2$  \_\_\_\_\_
5. CuBr \_\_\_\_\_
6.  $CO_2$  \_\_\_\_\_
7.  $PbSO_4$  \_\_\_\_\_
8.  $Li_2CO_3$  \_\_\_\_\_
9.  $Na_2CO_3$  \_\_\_\_\_
10.  $NO_2$  \_\_\_\_\_
11.  $N_2O_4$  \_\_\_\_\_
12.  $Ca(OH)_2$  \_\_\_\_\_
13.  $NH_4Cl$  \_\_\_\_\_
14.  $SO_3$  \_\_\_\_\_
15.  $AlPO_4$  \_\_\_\_\_
16.  $CCl_4$  \_\_\_\_\_
17. CdS \_\_\_\_\_
18.  $NH_3$  \_\_\_\_\_
19.  $MgI_2$  \_\_\_\_\_
20.  $K_3PO_4$  \_\_\_\_\_

Write the formulas for the following compounds.

1. carbon monoxide \_\_\_\_\_
2. sodium chloride \_\_\_\_\_
3. carbon tetrachloride \_\_\_\_\_
4. magnesium bromide \_\_\_\_\_
5. aluminum iodide \_\_\_\_\_
6. hydrogen hydroxide \_\_\_\_\_
7. iron (II) fluoride \_\_\_\_\_
8. carbon dioxide \_\_\_\_\_
9. sodium carbonate \_\_\_\_\_
10. ammonium sulfide \_\_\_\_\_
11. iron (II) oxide \_\_\_\_\_
12. iron (III) oxide \_\_\_\_\_
13. magnesium sulfate \_\_\_\_\_
14. sodium phosphate \_\_\_\_\_
15. dinitrogen pentoxide \_\_\_\_\_
16. phosphorus trichloride \_\_\_\_\_
17. aluminum sulfite \_\_\_\_\_
18. copper (I) carbonate \_\_\_\_\_
19. potassium hydrogen carbonate \_\_\_\_\_
20. sulfur trioxide \_\_\_\_\_