

Test 1 Practice—Fundamentals of Chemistry

Scientific Method

In chemistry class, Allen determined the effectiveness of various metals in releasing hydrogen gas from hydrochloric acid. Several weeks later, Allen read that a utilities company was burying lead next to iron pipes to prevent rusting. Allen conjectured that less rusting would occur with the more active metals. He placed the following into 4 separate beakers of water: (a) 1 iron nail, (b) 1 iron nail wrapped with an aluminum strip, (c) 1 iron nail wrapped with a magnesium strip, and (d) 1 iron nail wrapped with a lead strip. He used the same amount of water, equal amounts (mass) of the metals, and the same type of iron nails. At the end of 5 days, he rated the amount of rusting as small, moderate, or large. He also recorded the color of the water.

Identify the:

1. Problem
2. Hypothesis
3. Independent variable
4. Dependent variable
5. Control group
6. Observations
7. Constants

Scientific notation: Put the following in correct scientific notation

8. 56 000 000 000
9. 0.000 98
10. 0.198 765

Scientific notation: Answer each in correct scientific notation and with correct SF:

11. $(7.6 \times 10^{-11}) (6.1 \times 10^9)$
12. $\frac{3.58 \times 10^{-12}}{6.0 \times 10^8}$
13. $8.9 \times 10^7 - 2.1 \times 10^5$
14. $3.29 \times 10^4 + 1.21 \times 10^5$
15. $(9.8 \times 10^{-34}) (7 \times 10^{14})$

16. $\frac{5.6 \times 10^8}{3.19 \times 10^{12}}$

Significant figures: Determine the number of sig figs in the following numbers:

- 17. 608 cm
- 18. 200 kg
- 19. 0.007 00 m
- 20. 310.000 000 μg

Sig figs: Answer the following calculations with the correct units and sig figs:

- 21. $4.5\text{m} * 3.00\text{ m}$
- 22. $8.700\text{cm}/3.2\text{ cm}$
- 23. $7.80\text{ m} + 4\text{ m} + 78.2\text{ m}$
- 24. $0.64\text{ mm} - 4.3\text{ mm} - 0.200\text{ mm}$

Conversions (show all work with units!):

- 25. How many micrograms are in 45.6 kilograms?
- 26. How many meters are in 1050 cm?
- 27. Convert 35.38 mL to L.
- 28. How many inches are in 4.5×10^{-4} miles? (5280 ft = 1 mi)
- 29. Convert 50 km/hr to cm/s.
- 30. The speed limit on I-25 through Castle Rock is 65 mi/hour. Convert this to m/s. (1 mi = 1.61 km)
- 31. If I drive at 45 mi/hr, how many minutes will it take me to drive 60 miles?
- 32. The speed of light is 3.0×10^8 m/s. How many hours does it take light to travel 1.2×10^5 km?
- 33. How many seconds are in one century? (1 century = 100 years, 1 year = 365 days)
- 34. The earth has a volume of 1.08×10^{12} km³. How many cubic centimeters is this?
- 35. A bowling ball has a volume of 5300 cm³. Determine the volume in cubic meters.

Measurement: Practice measuring different objects with:

- 36. Graduated cylinders
- 37. Rulers
- 38. Triple beam balance