

Writing Formulas

More Practice

1. magnesium hydroxide
2. aluminum sulfate
3. barium hydroxide
4. diphosphorus pentoxide
5. nitric acid
6. sodium phosphate
7. copper (II) acetate
8. hydrochloric acid
9. carbon tetrachloride
10. sulfur hexafluoride
11. sulfur trioxide
12. ammonium sulfate
13. nickel (II) hydroxide
14. chromium (III) nitrate
15. carbon dioxide
16. magnesium phosphate
17. calcium oxide
18. aluminum sulfide
19. nitrous acid
20. ammonium nitride
21. ammonium nitrate
22. ammonium nitrite
23. sulfuric acid
24. potassium chloride
25. lithium carbonate
26. tin (IV) iodide
27. nickel (II) phosphate
28. barium nitrate
29. cobalt (II) sulfide
30. cobalt (II) sulfate
31. phosphorus pentoxide
32. sulfur trioxide
33. plumbous bromide plumbous = lead (II)
34. dinitrogen oxide
35. phosphorus trichloride
36. carbon tetrabromide
37. boron trichloride
38. carbonic acid
39. acetic acid
40. tetraiodine nonoxide
41. iron (III) oxalate oxalate = $C_2O_4^{2-}$
42. sodium phosphate
43. disulfur trioxide
44. magnesium nitride
45. calcium carbonate

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More Practice

1. magnesium hydroxide $Mg(OH)_2$
2. aluminum sulfate $Al_2(SO_4)_3$
3. barium hydroxide $Ba(OH)_2$
4. diphosphorus pentoxide P_2O_5
5. nitric acid (HNO_3)
6. sodium phosphate Na_3PO_4
7. copper (II) acetate $Cu(C_2H_3O_2)_2$
8. hydrochloric acid (HCl)
9. carbon tetrachloride CCl_4
10. sulfur hexafluoride SF_6
11. sulfur trioxide SO_3
12. ammonium sulfate $(NH_4)_2SO_4$
13. nickel (II) hydroxide $Ni(OH)_2$
14. chromium (III) nitrate $Cr(NO_3)_3$
15. carbon dioxide CO_2
16. magnesium phosphate $Mg_3(PO_4)_2$
17. calcium oxide CaO
18. aluminum sulfide Al_2S_3
19. nitrous acid (HNO_2)
20. ammonium nitride $(NH_4)_3N$
21. ammonium nitrate NH_4NO_3
22. ammonium nitrite NH_4NO_2
23. sulfuric acid H_2SO_4
24. potassium chloride KCl
25. lithium carbonate Li_2CO_3
26. tin (IV) iodide SnI_4
27. nickel (II) phosphate $Ni_3(PO_4)_2$
28. barium nitrate $Ba(NO_3)_2$
29. cobalt (II) sulfide ~~CoS~~ CoS
30. cobalt (II) sulfate $CoSO_4$
31. phosphorus pentoxide PO_5
32. sulfur trioxide SO_3
33. plumbous bromide $PbBr_2$
34. dinitrogen oxide N_2O
35. phosphorus trichloride PCl_3
36. carbon tetrabromide CBr_4
37. boron trichloride BCl_3
38. carbonic acid H_2CO_3
39. acetic acid $H(C_2H_3O_2)$
40. tetraiodine nonoxide I_4O_9
41. iron (III) oxalate $Fe_2(C_2O_4)_3$
42. sodium phosphate Na_3PO_4
43. disulfur trioxide S_2O_3
44. magnesium nitride Mg_3N_2
45. calcium carbonate $CaCO_3$